



Guide to Best Practices in Ocean Acidification Research and Data Reporting

Addendum, 2015-11-13

This addendum provides critical information and links to complementary resources for users of the *Guide to Best Practices in Ocean Acidification Research and Data Reporting* (Eds: Riebesell U., Fabry V. J., Hansson L. & Gattuso J.-P., 2010. 260 p. Luxembourg: Publications Office of the European Union). It was initiated following discussions within the SOLAS IMBER Ocean Acidification Working Group and the Advisory Board of the IAEA Ocean Acidification International Coordination Centre.

Contributors: Jean-Pierre Gattuso, Andrew Dickson, Sam Dupont, Jens Nejstgaard and Ulf Riebesell.

The following references, listed according to Chapter, provide important additional information on the various aspects discussed in the Best Practices guide. Users are strongly recommended to consult these articles as a complement to the Guide.

Chapter 1: The carbon dioxide system in seawater: equilibrium chemistry and measurements, by A. G. Dickson

Bockmon, E. E., & Dickson, A. G. 2014. A seawater filtration method suitable for total dissolved carbon and pH analyses. *Limnology and Oceanography: Methods* 12, 191–195. <http://www.aslo.org/lomethods/free/2014/0191.html>.

Bockmon, E. E., & Dickson, A. G. 2015. An inter-laboratory comparison assessing the quality of seawater carbon dioxide measurements. *Marine Chemistry* 171, 36-43. <http://www.sciencedirect.com/science/article/pii/S0304420315000213>

Hoppe C. J. M., Langer G., Rokitta S. D., Wolf-Gladrow D. A. & Rost B., 2012. Implications of observed inconsistencies in carbonate chemistry measurements for ocean acidification studies. *Biogeosciences* 9:2401-2405. <http://www.biogeosciences.net/9/2401/2012/bg-9-2401-2012.html>.

Koeve W., Kim H.-C., Lee K. & Oschlies A., 2012. Potential impact of DOC accumulation on fCO₂ and carbonate ion computations in ocean acidification experiments. *Biogeosciences* 9:3787-3798. <http://www.biogeosciences.net/9/3787/2012/bg-9-3787-2012.html>.

- Martz T.R., Daly K. L., Byrne R. H., Stillman J. H. & Turk D., 2015. Technology for ocean acidification research: Needs and availability. *Oceanography* 28(2):40–47. http://www.tos.org/oceanography/archive/28-2_martz.html

Orr J. C., Epitalon J.-M. & Gattuso J.-P., 2015. Comparison of ten packages that compute ocean carbonate chemistry. *Biogeosciences* 12:1483-1510. <http://www.biogeosciences.net/12/1483/2015/bg-12-1483-2015.html>.

See also Appendix 1 to this Addendum: “*Guidelines for reporting ocean acidification data in scientific journals*”

Chapter 2: Approaches and tools to manipulate the carbonate chemistry, by J.-P. Gattuso et al.

Bockmon E. E., Frieder C. A., Navarro M. O., White-Kershek L. A. & Dickson A. G., 2013. Technical note: controlled experimental aquarium system for multi-stressor investigation of carbonate

chemistry, oxygen saturation, and temperature. *Biogeosciences* 10:5967-5975. <http://www.biogeosciences.net/10/5967/2013/bg-10-5967-2013.html>

Cornwall C. E. & Hurd C. L., in press. Experimental design in ocean acidification research: problems and solutions. *ICES Journal of Marine Science*. <http://icesjms.oxfordjournals.org/content/early/2015/07/07/icesjms.fsv118.abstract>

Hoffmann L. J., Breitbarth E., McGraw C. M., Law C. S., Currie K. I. & Hunter K. A., 2013. A trace-metal clean, pH-controlled incubator system for ocean acidification incubation studies. *Limnology and Oceanography Methods* 11:53-61. <http://www.aslo.org/lomethods/free/2013/0053.html>

Hoppe C. J. M., Langer G., Rokitta S. D., Wolf-Gladrow D. A. & Rost B., 2012. Implications of observed inconsistencies in carbonate chemistry measurements for ocean acidification studies. *Biogeosciences* 9:2401-2405. <http://www.biogeosciences.net/9/2401/2012/bg-9-2401-2012.html>.

Jokiel P. L., Bahr K. D. & Rodgers K. S., 2014. Low-cost, high-flow mesocosm system for simulating ocean acidification with CO₂ gas. *Limnology & Oceanography Methods* 12(5):313-322. <http://onlinelibrary.wiley.com/doi/10.4319/lom.2014.12.313/abstract>

Koeve W., Kim H.-C., Lee K. & Oschlies A., 2012. Potential impact of DOC accumulation on fCO₂ and carbonate ion computations in ocean acidification experiments. *Biogeosciences* 9:3787-3798. <http://www.biogeosciences.net/9/3787/2012/bg-9-3787-2012.html>

Lunden J. J., Turner J. M., McNicholl C. G., Glynn C. K. & Cordes E. E., 2014. Design, development, and implementation of recirculating aquaria for maintenance and experimentation of deep-sea corals and associated fauna. *Limnology and Oceanography: Methods* 12(6):363-372. <http://onlinelibrary.wiley.com/doi/10.4319/lom.2014.12.363/abstract>

MacLeod C. D., Doyle H. L. & Currie K. I., 2015. Technical Note: Maximising accuracy and minimising cost of a potentiometrically regulated ocean acidification simulation system. *Biogeosciences* 12:713-721. <http://www.biogeosciences.net/12/713/2015/bg-12-713-2015.html>

Olariaga A., Guallart E. F., Fuentes V., López-Sanz A., Canepa A., Movilla J., Bosch M., Calvo E. & Pelejero C., 2014. Polyp flats, a new system for experimenting with jellyfish polyps, with insights into the effects of ocean acidification. *Limnology & Oceanography Methods* 12:212-222. <http://www.aslo.org/lomethods/free/2014/0212.html>

Torres R., Manriquez P. H., Duarte C., Navarro J. M., Lagos N. A., Vargas C. A. & Lardies M. A., 2013. Evaluation of a semi-automatic system for long-term seawater carbonate chemistry manipulation. *Revista Chilena de Historia Natural* 86(4):443-451. http://www.scielo.cl/scielo.php?script=sci_arttext&pid=S0716-078X2013000400006

Wilcox-Freeburg E., Rhyne A., Robinson W. E., Tlusty M., Bourque B. & Hannigan R. E., 2013. A comparison of two pH-stat carbon dioxide dosing systems for ocean acidification experiments. *Limnology and Oceanography: Methods* 11:485-494. <http://onlinelibrary.wiley.com/doi/10.4319/lom.2013.11.485/abstract>.

Chapter 3: Atmospheric CO₂ targets for ocean acidification perturbation experiments, by J. P. Barry et al.

Cornwall C. E. & Hurd C. L., in press. Experimental design in ocean acidification research: problems and solutions. *ICES Journal of Marine Science*. <http://icesjms.oxfordjournals.org/content/early/2015/07/07/icesjms.fsv118.abstract>

Reum J. C. P., Alin S. R., Harvey C. J., Bednaršek N., Evans W., Feely R. A., Hales B., Lucey N., Mathis J. T., McElhany P., Newton L. & Sabine C. L., in press. Interpretation and design of ocean acidification experiments in upwelling systems in the context of carbonate chemistry co-variation with temperature and oxygen. *ICES Journal of Marine Science*. <http://icesjms.oxfordjournals.org/content/early/2015/01/06/icesjms.fsu231.abstract>

General book on experimental design:

Quinn G. & Keough M., 2002. Experimental design and data analysis for biologists. *Cambridge University Press*, <http://www.zoology.unimelb.edu.au/qkstats/>

Wahl M., Sawall Y. & Saderne V., in press. How good are we at assessing the impact of ocean acidification in coastal systems? Limitations, omissions and strengths of commonly used experimental approaches with a special emphasis on the neglected role of fluctuations. *Marine & Freshwater Research*. http://www.publish.csiro.au/view/journals/dsp_journals_pip_abstract_Scholar1.cfm?nid=126&pip=M F14154

Chapter 4:

Anderson D. R., Burnham K. P. & Thompson W. L., 2000. Null hypothesis testing: problems, prevalence, and an alternative. *The journal of wildlife management* 64(4):912-923. http://www.jstor.org/stable/3803199?seq=1#page_scan_tab_contents

Cornwall C. E. & Hurd C. L., in press. Experimental design in ocean acidification research: problems and solutions. *ICES Journal of Marine Science*. <http://icesjms.oxfordjournals.org/content/early/2015/07/07/icesjms.fsv118.abstract>

Ellison A. M., Gotelli N. J., Inouye B. D. & Strong D. R., 2014. P values, hypothesis testing, and model selection: it's déjà vu all over again 1. *Ecology* 95:609-610. <http://www.esajournals.org/doi/abs/10.1890/13-1911.1>

Masson M. E., 2011. A tutorial on a practical Bayesian alternative to null-hypothesis significance testing. *Behavior Research Methods* 43:679-690. <http://rd.springer.com/article/10.3758%2Fs13428-010-0049-5>

Smith R. A., Levine T. R., Lachlan K. A. & Fediuk T. A., 2002. The high cost of complexity in experimental design and data analysis: Type I and type II error rates in multiway ANOVA. *Human Communication Research* 28:515-530. <http://onlinelibrary.wiley.com/doi/10.1111/j.1468-2958.2002.tb00821.x/abstract>

Veresoglou S. D., 2015. P hacking in biology: An open secret. *Proceedings of the National Academy of Sciences of the United States of America (PNAS)* 112(37):E5112-E5113. <http://www.pnas.org/content/112/37/E5112.full.pdf>

Wahl M., Sawall Y. & Saderne, V., in press.. How good are we at assessing the impact of ocean acidification in coastal systems? Limitations, omissions and strengths of commonly used experimental approaches with a special emphasis on the neglected role of fluctuations. *Marine and Freshwater Research*. <http://www.publish.csiro.au/paper/MF14154.htm>

Chapter 6: Pelagic mesocosms, by U. Riebesell et al.

Czerny J., Schulz K. G., Boxhammer T., Bellerby R. G. J., Büdenbende, J., Engel A., Krug S. A., Ludwig A., Nachtigall K., Nondal G., Niehoff B., Silyakova A. & Riebesell U., 2013. Implications of elevated CO₂ on pelagic carbon fluxes in an Arctic mesocosm study – an elemental mass balance approach. *Biogeosciences* 10(5):3109–3125. <http://www.biogeosciences.net/10/3109/2013/bg-10-3109-2013.html>

de Kluijver A., Soetaert K., Czerny J., Schulz K. G., Boxhammer T., Riebesell U. & Middelburg J. J., 2013. A ¹³C labelling study on carbon fluxes in Arctic plankton communities under elevated CO₂ levels. *Biogeosciences* 10(3):1425–1440. <http://www.biogeosciences.net/10/1425/2013/bg-10-1425-2013.html>

Guieu C., Dula F., Desboeufs K., Wagener T., Pulido-Villena E., Grisoni J. M., Louis F., Ridame C., Blain S., Brunet C., Bon Nguyen E., Tran S., Labiadh M. & Dominici J. M., 2010. Large clean mesocosms and simulated dust deposition: a new methodology to investigate responses of marine oligotrophic ecosystems to atmospheric inputs. *Biogeosciences* 7(9):2765–2784. <http://www.biogeosciences.net/7/2765/2010/bg-7-2765-2010.html>

Kangas P. & Adey W., 1996. Mesocosms and ecological engineering. *Ecological Engineering* 6:1-5. <https://www.infona.pl/resource/bwmetal.element.elsevier-10c3b6bb-76de-3fdd-b3e2-90cee0aee248/tab/summary>

Larsen A., Egge J. K., Nejtgaard J. C., Di Capua I., Thyraug R., Bratbak G. & Thingstad T. F., 2015. Contrasting response to nutrient manipulation in Arctic mesocosms are reproduced by a minimum microbial food web model. *Limnology and Oceanography* 60:360-374. <http://onlinelibrary.wiley.com/doi/10.1002/lno.10025/abstract>

Petersen J. E., Kennedy V. S., Dennison W. C. & Kemp W. M. (Eds), 2009. *Enclosed Experimental Ecosystems and Scale. Tools for Understanding and Managing Coastal Ecosystems*. Springer, New York, NY, USA. <http://www.springer.com/us/book/9780387767666>

Riebesell U., Czerny J., von Bröckel K., Boxhammer T., Büdenbender J., Deckelnick M., Fischer M., Hoffmann D., Krug S. A., Lentz U., Ludwig A., Muche R. & Schulz K. G., 2013. Technical Note: A mobile sea-going mesocosm system – new opportunities for ocean change research. *Biogeosciences* 10(3):1835–1847. <http://www.biogeosciences.net/10/1835/2013/bg-10-1835-2013.html>

Thingstad T. F. & Cuevas L. A., 2010. Nutrient pathways through the microbial food web: principles and predictability discussed, based on five different experiments. *Aquatic Microbial Ecology* 61:249–260. <http://www.int-res.com/abstracts/ame/v61/n3/p249-260/>

Vadstein O., Andersen T., Reinertsen H. R. & Olsen Y., 2012. Carbon, nitrogen and phosphorus resource supply and utilisation for coastal planktonic heterotrophic bacteria in a gradient of nutrient loading. *Marine Ecology Progress Series* 447:55–75. <http://www.int-res.com/abstracts/meps/v447/p55-75/>

Chapter 8: In situ perturbation experiments: natural venting sites, spatial/temporal gradients in ocean pH, manipulative in situ $p(\text{CO}_2)$ perturbations, by J. P. Barry et al.

Campbell J. E. & Fourqurean J. W., 2011. Novel methodology for in situ carbon dioxide enrichment of benthic ecosystems. *Limnology and Oceanography: Methods* 9:97-109. <http://www.aslo.org/lomethods/free/2011/0097.html>

Czerny J., Schulz K. G., Ludwig A. & Riebesell U., 2013. Technical note: a simple method for air–sea gas exchange measurements in mesocosms and its application in carbon budgeting. *Biogeosciences* 10:1379-1390. <http://www.biogeosciences.net/10/1379/2013/bg-10-1379-2013.html>

Gattuso J.-P., Kirkwood W., Barry J. P., Cox E., Gazeau F., Hansson L., Hendriks I. E., Kline D. I., Mahacek P., Marker M., Martin S., McElhany P., Peltzer E. T., Reeve J., Roberts D., Saderne V., Tait K., Widdicombe S. & Brewer P., 2014. Free-ocean CO_2 enrichment (FOCE) systems: present status and future developments. *Biogeosciences* 11:4057-4075. <http://www.biogeosciences.net/11/4057/2014/bg-11-4057-2014.html>

Kirkwood W. J., Walz P. M., Peltzer E. T., Barry J. P., Herlien R. A., Headley K. L., Key C., Matsumoto G. I., Maughan T., O'Reilly T. C., Salamy K. A., Shane F. & Brewer P. G., 2015. Design, construction, and operation of an actively controlled deep-sea CO_2 enrichment experiment using a cabled observatory system. *Deep Sea Research Part I: Oceanographic Research Papers* 97:1-9. <http://www.sciencedirect.com/science/article/pii/S096706371400209X>

Riebesell U., Czerny J., von Bröckel K., Boxhammer T., Büdenbender J., Deckelnick M., Fischer M., Hoffmann D., Krug S. A., Lentz U., Ludwig A., Mucche R. & Schulz K. G., 2013. Technical Note: A mobile sea-going mesocosm system – new opportunities for ocean change research. *Biogeosciences* 10:1835-1847. <http://www.biogeosciences.net/10/1835/2013/bg-10-1835-2013.html>

Annex 1

Guidelines for reporting ocean acidification data in scientific journals

These guidelines were prepared by J.-P. Gattuso (gattuso@obs-vlfr.fr), H. Garcia (hernan.garcia@noaa.gov), C. J. M. Hoppe (Clara.Hoppe@awi.de), J. Orr (James.Orr@lsce.ipsl.fr), H.-O. Pörtner (Hans.Poertner@awi.de) and Y Yang (yangyan@xmu.edu.cn)

This document was prepared in the framework of the data management activity of the Ocean Acidification International Coordination Centre of the International Atomic Energy Agency (OAICC; www.iaea.org/ocean-acidification). Please contact the first author (gattuso@obs-vlfr.fr) in case of any error or omission. It is primarily based on Dickson et al. (2007), Dickson (2010), Nisumaa et al. (2010), Pesant et al. (2010), Pörtner et al. (2010) and Orr et al. (2015).

To ensure reproducibility, it is critical to report at least two variables of the carbonate system of seawater as well as salinity, temperature, and the hydrostatic pressure (if the measurements were not performed at atmospheric pressure). In addition, authors should report concentrations of total dissolved inorganic phosphorus as well as total dissolved inorganic silicon (in $\mu\text{mol kg}^{-1}$) whenever possible. Furthermore,

- Authors should carefully report how the parameters were measured and, if applicable, which protocol they followed.
- The use of Certified Reference Materials, source, and batch numbers must be mentioned.
- At least two of the following carbonate system parameters should be measured and reported (note the preferred acronyms and units):
 - Dissolved inorganic carbon (C_T ; $\mu\text{mol kg}^{-1}$)
 - Total alkalinity (A_T ; $\mu\text{mol kg}^{-1}$)
 - pH (it is critical to mention its scale; see below)
 - Partial pressure of carbon dioxide ($p\text{CO}_2$; μatm)
 - Fugacity of carbon dioxide ($f\text{CO}_2$; μatm)
 - Carbonate ion concentration (CO_3^{2-} ; $\mu\text{mol kg}^{-1}$)
- The pH scale (NBS, free, total, or seawater) must be mentioned prominently in the manuscript.
- If more than one pH scale is used in a given manuscript, the pH should always be given with the associated scale as a subscript:
 - on the National Bureau of Standards scale (pH_{NBS})
 - on the seawater scale (pH_{SWS})

- on the free scale (pH_F)
- on the total scale (pH_T)
- The temperature at the time of sampling and at the time of measurement should both be mentioned, if they differ.
- Salinity is needed (note that it is unitless).
- The formulations used to calculate the following variables should be mentioned:
 - Concentrations of total boron
 - CO_2 solubility (K_0)
 - Dissociation constants of carbonic acid (K_1 and K_2), boric acid (K_b), water (K_w), phosphoric acid (K_{p1} , K_{p2} , K_{p3}), silicic acid (K_{Si}), hydrogen fluoride (K_f), and bisulfate (K_s)
 - Solubility products of calcite (K_{spc}) and aragonite (K_{spa})
- The software package used to calculate the carbonate chemistry, along with its version number, and any associated options must all be mentioned.
- Average reproducibility of the performed measurements (with number of measurements) should be mentioned.
- Finally, it is strongly recommended that the chemistry and biological data are either archived in an on-line database (preferred) or provided along with the paper as supplementary information.

References cited

- Dickson, A. G.: The carbon dioxide system in seawater: equilibrium chemistry and measurements, in: Guide to best practices for ocean acidification research and data reporting, edited by: Riebesell, U., Fabry, V. J., Hansson, L., and Gattuso, J.-P., Publications Office of the European Union, Luxembourg, 17–40, 2010.
- Dickson, A. G., Sabine, C. L., and Christian J. R.: Guide to best practices for ocean CO_2 measurements, PICES Special Publication, 3, 1–191, 2007.
- Nisumaa, A.-M., Pesant, S., Bellerby, R. G. J., Delille, B., Middelburg, J. J., Orr, J. C., Riebesell, U., Tyrrell, T., Wolf-Gladrow, D., and Gattuso, J.-P.: EPOCA/EUR-OCEANS data compilation on the biological and biogeochemical responses to ocean acidification, Earth Syst. Sci. Data, 2, 167–175, doi:10.5194/essd-2-167-2010, 2010.
- Orr J. C., Epitalon J.-M., and Gattuso J.-P., 2015. Comparison of ten packages that compute ocean carbonate chemistry, Biogeosciences, 12, 1483–1510, 2015.

Pesant, S., Alan Hook, L., Lowry, R., Moncoiffé, G., Nisumaa, A.-M., and Pfeil, B.: Safeguarding and sharing ocean acidification data, in: Guide to best practices for ocean acidification research and data reporting, edited by: Riebesell, U., Fabry, V. J., Hansson, L., and Gattuso, J.-P., Publications Office of the European Union, Luxembourg, 243–258, 2010.

Pörtner H.-O., Dickson A., and Gattuso J.-P.: Terminology and units for parameters relevant to the carbonate system, in: Guide to best practices for ocean acidification research and data reporting, edited by: Riebesell, U., Fabry, V. J., Hansson, L., and Gattuso, J.-P., Publications Office of the European Union, Luxembourg, 18–19, 2010.